



"VASILE ALECSANDRI" UNIVERSITY OF BACĂU, ROMANIA
ENGINEERING FACULTY

"GHEORGHE ASACHI" UNIVERSITY OF IASI, ROMANIA
POLITEHNICA UNIVERSITY OF BUCURESTI, ROMANIA



ROMANIAN ACADEMY OF AGRICULTURAL AND FORESTRY SCIENCES "GHEORGHE IONESCU – SISESTI"

POLITECNICO UNIVERSITY OF TORINO, ITALY

"ANGEL KANCHEV" UNIVERSITY OF RUSE, BULGARIA

WARSAW UNIVERSITY OF TECHNOLOGY, POLAND

ROMANIAN ASSOCIATION OF MANAGERS AND ECONOMIC ENGINEERS



THE INTERNATIONAL CONFERENCES

**CELEBRATING 55 YEARS OF HIGHER EDUCATION
AND 40 YEARS OF TECHNICAL HIGHER
EDUCATION INTO**

**"VASILE ALECSANDRI" UNIVERSITY OF BACAU
OPROTEH-2016**

CIEI-2016

CISA- 2016

CIER-2016

**The 22th National Symposium of Fracture Mechanics
2016**

**"Moise Tuturea" Students Contest Society on
Engineering and Management**

Conference Proceedings

ABSTRACTS

BACĂU

JUNE 02-04, 2016

3.A.8. COMPARATIVE ADSORPTION STUDIES OF TWO ANIONIC DYES FROM AQUEOUS SOLUTION USING LAYERED DOUBLE HYDROXIDES

M. C. ALEXANDRICA, M. SILION, G. DODI, D. NISTOR, R. D. ANDREI, M. I. POPA

"Gheorghe Asachi" Technical University of Iasi, 71D, Mangeron, 700050, Romania

"Petru Poni" Institute of Macromolecular Chemistry, 41 A, Aleea Grigore Ghica Vodă, 700487, Iasi, Romania

SCIENT - Research Centre for Instrumental Analysis, Bucharest, Romania

"Vasile Alecsandri" University of Bacau, Calea Marasesci 156, Bacau, 600115, Romania

"Charles Gerhardt" Institute, UMR 5253, Laboratoire de Matériaux Avancés pour la Catalyse et de la Santé, Montpellier, France

Layered double hydroxides (LDH) with hydrotalcite-like structure containing Mg(II) and Al(III) in the layers were prepared at different Mg/Al molar ratio by co-precipitation method at constant pH (denoted MgALDH). The prepared LDH was used for Alizarin Orange (AO) and Orange G (OG) dyes removal from aqueous solutions. Sorption batch experiments were conducted for optimizing the pH, initial dye concentration, sorbent dose, contact time and temperature. Physicochemical characterization of MgALDH before and after sorption studies was carried out by XRD, FTIR, EDX, SEM and BET analysis. The sorption kinetics was analyzed by fitting the data into the following models: pseudo-first-order, pseudo-second-order and intraparticle diffusion. The Langmuir, Freundlich and Dubinin-Radushkevich models were used to fit the adsorption isotherms. The results indicated that Mg/Al layered double hydroxide is an efficient adsorbent for the treatment of wastewater with different concentrations of anionic dyes.

Keywords: MgALDH, anionic dyes, adsorption studies, kinetic.

3.A.9. DEVELOPMENT OF POLYMER HYBRID COMPOSITE FILMS WITH TUNABLE WETTABILITY BY SURFACE MODIFICATIONS

G. DODI, D. HRITCU, D. DRAGANESCU, V. CARLESCU, M. L. IORDACHE, M. I. POPA

"Gheorghe Asachi" Technical University of Iasi, Faculty of Chemical Engineering and Environmental Protection, Romania

"Gheorghe Asachi" Technical University of Iasi, Department of Mechanical Engineering, Mechatronics and Robotics, Romania

meses@it.iasi.ro

The objective of the present study was to evaluate and compare the wetting behavior of hybrid composite coatings prepared with two different UV initiators for water-repellent applications. The effect of coating chemistry was studied in order to obtain micro hierarchical surface roughness. The coatings were applied in a layer-by-layer configuration: hybrid polymeric matrix as a base layer, followed by nanoparticle dispersion as the top layer. Various polymeric matrices were tested, namely chitosan bearing surface vinyl groups, subsequently cross-linked with ethylene glycol dimethacrylate (EGDMA) using UV initiation at specific wavelengths. (2-Hydroxy-2-methylpropiophenone (HMP) and benzoin methyl ether (BMM) were used as UV initiators for radical polymerization of modified chitosan with EGDMA. The effects of polymer, monomer and initiators concentration and also the solvent volume

Acknowledgement: M. Silon is grateful for the financial support from the H2020 ERA Chairs Project no. 667387, SupraChem Lab Laboratory of Supramolecular Chemistry for Adaptive Delivery Systems ERA Chair initiative and the Romanian National Authority for Scientific Research, CNCS - UEFISCDI, project number PN-III-PCE-2014-2-0029.

Keywords: layered double hydroxides, tramadol hydrochloride, intercalation, controlled release

Tramadol hydrochloride (TH) is an opioid pain medication used to treat moderate to moderately severe pain and when taken as an immediate-release oral formulation, the onset of pain relief usually occurs within about an hour. For this reason was intercalated into Zn-Al-layered double hydroxides (ZnALDHs) by ion exchange method for controlled drug delivery. LDHs matrix based on Zn(II) and Al(III) at different molar ratios were prepared by co-precipitation method at constant pH, followed by TH solution addition in basic medium for the formation of LDH-drug systems. The samples were characterized by X-ray diffraction (XRD), Fourier transform infrared (FTIR), Raman spectroscopy, thermogravimetric analysis (TG) and scanning electron microscopy (SEM). The analysis confirmed the successful intercalation of TH anions into the interlayer of LDHs. Release of the intercalated TH was studied in simulated gastric fluid (pH = 2) and simulated intestinal fluid (pH = 7.4) at 37 °C. Controlled release of TH from ZnALDH/TH was observed. This feature proves that these hybrid materials can be classified as controlled release formulations for the TH under study.

George Emil Asachi Technical University of Iasi, 71D, Mangeron, 700050, Romania
Petru Poni Institute of Macromolecular Chemistry, 41 A, Aleea Eroilor, 700487, Iasi, Romania

M. C. ALEXANDRICA, M. SILON, M. I. POPA

3.1.10. LAYERED DOUBLE HYDROXIDES AS EFFECTIVE CARRIER FOR THE INTERCALATION OF TRAMADOL HYDROCHLORIDE

Acknowledgement: This work was supported by a grant of the Romanian Ministry of National Education, CNCS-UEFISCDI, project number PN-III-PCE-2012-4-0433.

used for the polymerization were studied in order to yield coatings with high wetting angle. Also, the influence of the coating structure, that is, the addition of iron oxide nanoparticles capable of magnetic self-assembly during the curing stage and pre-hybridized/precondensed sol-gel solution of hexadecyltrimethoxy silane as a coupling agent, incorporated both into the matrix and the nanoparticle suspension, was evaluated. The magnetic nanoparticles were prepared by mild oxidation of ferrous ions in alkaline solution, followed by ammonia with (3-aminopropyl) triethoxysilane. Hybrid nanoparticle-polymer films prepared by spraying were deposited and cured by drying on glass slides. The water contact angle measurements evidenced hydrophobic surfaces for the newly prepared hybrid films.